

Solvolytic Rearrangement of 5,6-Polymethylenebicyclo[4.2.0]octan-2-yl  
Derivatives to Functionalized Polymethylenebicyclo[3.2.1]- and [3.3.0]octanes

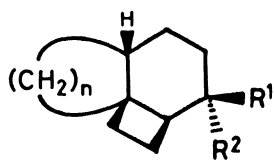
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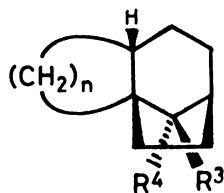
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The functionalized polymethylenebicyclo[3.2.1]- and [3.3.0]octane derivatives are newly synthesized from endo- and exo-5,6-polymethylenebicyclo[4.2.0]oct-2-yl 3,5-dinitrobenzoates by skeletal rearrangement using solvolysis.

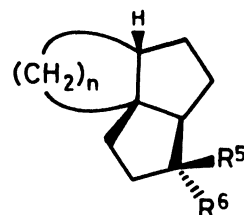
Recently, much attention has been paid on the natural products having bicyclo[3.2.1]- or [3.3.0]octane ring system as a key structure because of their unique carbon skeletons and biological activities.<sup>1)</sup> We have studied on the facile construction of the basic frameworks of polyquinane sesquiterpenes by the acid-catalyzed rearrangement of trimethylenebicyclo[4.2.0]octane derivatives.<sup>2)</sup> In this connection, we describe herein novel construction of the functionalized bicyclo[3.2.1]- and [3.3.0]octane ring systems by solvolytic rearrangement of the endo- and exo-cis,cis-5,6-polymethylenebicyclo[4.2.0]oct-2-yl 3,5-dinitrobenzoates 1a-c and 2a-c.



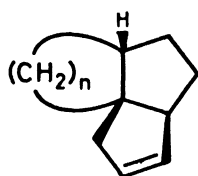
1a, b, c  $R^1 = \text{ODNB}, R^2 = \text{H}$   
2a, b, c  $R^1 = \text{H}, R^2 = \text{ODNB}$   
3a  $R^1 = \text{OCO} \text{ (benzene ring) } \text{Br}, R^2 = \text{H}$



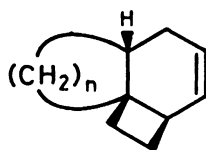
4a, b, c  $R^3 = \text{H}, R^4 = \text{ODNB}$   
5a, b, c  $R^3 = \text{H}, R^4 = \text{OH}$   
6a, b, c  $R^3 = \text{OH}, R^4 = \text{H}$



7a, b, c  $R^5 = \text{ODNB}, R^6 = \text{H}$   
8a, b, c  $R^5 = \text{OH}, R^6 = \text{H}$   
9a, b, c  $R^5 = \text{H}, R^6 = \text{OH}$



10a, b, c



11a, b, c

a: n=3, b: n=4, c: n=5  
ODNB= 3,5-dinitrobenzoyl

The endo- and exo-benzoates 1 and 2 were easily prepared from the corresponding alcohols,<sup>4)</sup> and their stereochemistry was determined on the basis of our previous work<sup>5)</sup> and the X-ray analysis of endo-p-bromobenzoate 3a.<sup>6)</sup>

Solvolytic rearrangements of 1a-2c were carried out in 50% (v/v) aqueous dioxane at 160 °C except for 2a at 150 °C. As for the rates of hydrolysis,<sup>7)</sup> no marked difference was observed between 1 and 2. The product distribution is shown in Table 1.<sup>8)</sup> The endo-benzoates 1a-c gave the bicyclo[3.3.0]octane derivatives 7-10 mainly. Although the content of unidentified olefins increased in the case of higher homologues, the exo-benzoates 2a-c gave the bicyclo[3.2.1]octane derivatives 4-6 predominantly. This significantly high selectivity is explained on the ground of the principle of stereoelectronic control. Namely, in the series of endo-benzoates (1a-c), the central bond of the bicyclo[4.2.0]octane unit migrates preferentially, whereas in the case of exo-benzoates (2a-c) migration of the peripheral bond predominates because of the antiperiplanar alignment between the migrating bond and the leaving group.

Table 1. Product distribution for hydrolysis of 3,5-dinitrobenzoates 1 and 2

Benzoate	Products/%									Yield/%
	<u>4</u>	<u>5</u>	<u>6</u>	<u>7</u>	<u>8</u>	<u>9</u>	<u>10</u>	<u>11</u>	Others <sup>a)</sup>	
<u>1a</u>	-	2	-	19	44	17	10	5	3	58 <sup>b)</sup>
<u>1b</u>	-	4	22	24	8	27	9	5	1	92
<u>1c</u>	-	3	-	34	20	16	10	3	14	93
<u>2a</u>	56	29	6	-	-	-	-	5	4	84
<u>2b</u>	48	30	1	-	-	1	-	5	15	82
<u>2c</u>	29	19	-	-	2	3	-	12	35	90

a) Unidentified olefins.

b) A considerable amount of polymeric materials was obtained.

## References

- 1) For recent reviews, see: L. A. Paquette, *Top. Curr. Chem.*, **119**, 1 (1984); E. Fujita and M. Node, *Progress in the Chemistry of Natural Product*, **46**, 77(1984).
- 2) Y. Odaira and K. Kakiuchi, *Yuki Gosei Kagaku Kyokai Shi*, **44**, 930 (1986).
- 3) All new compounds gave satisfactory spectral and analytical data.
- 4) Prepared by photocycloaddition of the corresponding bicyclic enones to ethylene followed by hydride reduction; see: M. Ue, K. Kobiro, K. Kakiuchi, Y. Tobe, and Y. Odaira, *Chem. Lett.*, **1987**, 1903.
- 5) K. Kakiuchi, S. Kumanoya, M. Ue, Y. Tobe, and Y. Odaira, *Chem. Lett.*, **1985**, 989.
- 6) Details will be reported elsewhere.
- 7) The rates of hydrolysis were measured by titrimetric method. The rate constants (50% aqueous dioxane at 150 °C) are as follows:  
 $\underline{1a}$ :  $6.99 \times 10^{-6} \text{ s}^{-1}$ ;  $\underline{1b}$ :  $3.28 \times 10^{-5} \text{ s}^{-1}$ ;  $\underline{1c}$ :  $1.62 \times 10^{-5} \text{ s}^{-1}$ .  
 $\underline{2a}$ :  $1.49 \times 10^{-5} \text{ s}^{-1}$ ;  $\underline{2b}$ :  $9.98 \times 10^{-5} \text{ s}^{-1}$ ;  $\underline{2c}$ :  $1.29 \times 10^{-5} \text{ s}^{-1}$ .
- 8) The structure of solvolytic products were confirmed by transformation to suitable derivatives of which authentic samples were prepared independently.

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